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THIRD SEMESTER M.Sc. DEGREE (REGULAR) EXAMINATION NOVEMBER 2019

Chemistry

CH 3C 10-ORGANOMETALIC AND BIO-INORGANIC CHEMISTRY

Time: Three Hours

Maximum: 36 Weightage

Part A

Answer all questions.

Each question carries a weightage of 1.

- 1. What is 18 electron rule? Check whether the complex η^5 -(C₅H₅) Fe (CO)₂ Cl obeys this rule.
- 2. Metals which are in high oxidation state or surrounded by strong π acceptor ligands can form stable dihydrogen complexes. Why?
- 3. What is hapticity of an organic ligand? Predict the hapticity of cyclopentadienyl ligand in ferrocene.
- 4. Explain the synergistic effect of CO in metal carbonyls.
- 5. What is Zeise's salt? The C-C bond length in zeise's salt is longer than in free ethylene. Why?
- 6. Ir spectroscopy is a very useful technique to study the progress of ligand substitution reactions in metal carbonyls. Explain.
- 7. What is Zeigler-Natta catalyst? Early transition metal halides can function as good Zeigler-Natta catalyst along with aluminium alkyls, but not late transition metal halides. Why?
- 8. What is hydroformylation reaction? Which is the catalyst used? Give the industrial importance of this reaction.
- 9. Simple heme units cannot act as oxygen carriers. Why?
- 10. What is haemocyanin? What is its role in living system? Its oxy form is blue and deoxy form is colourless. Why?
- 11. Why Co-based macrocyclic complex is well suited for radical-based rearrangements rather than Fe-complex like haem?
- 12. What is superoxide dismutase? Which is the active site in the enzyme? What is the role of this enzyme?

 $(12 \times 1 = 12 \text{ weightage})$

Turn over

Part B

6

Answer any eight questions.

Each question carries a weightage of 2.

- 13. What is Collman's reagent? How it is prepared? Discuss its significance in synthetic organic chemistry.
- 14. Determine the total electron count, polyhedral electron count and predicted structure of the following metal clusters. (1) $[Ru_6N(CO)_{16}]^-$, (2) $Os_5(CO)_{16}$, (3) $Ru_5C(CO)_{16}$ (4) $Rh_6C(CO)_{16}$.
- 15. Discuss olefin metathesis reaction with example.
- 16. Write a note on substitution reactions in organometallic complexes. With suitable examples differentiate between associative and dissociative substitution reactions.
- 17. What type of reaction is the following, insertion or migration? Justify your answer. $(CH_3)Mn(CO)_5 + CO \rightarrow (CH_3CO)Mn(CO)_5$.
- 18. Give one method of preparation of cyclobutadiene complex. Discuss the bonding. Cyclobutadiene is more stable in the complex form than free ligand. Why?
- 19. What are chevrel compounds? Give their method of preparation and structure. Discuss their important applications.
- 20. What is isolobal analogy? Justifythe isolobality of the following species. H, Cl, CH₃ Co(CO)₄, Fe(Cp)(CO)₂.
- 21. Briefly explain the mode of transport and storage of iron in living organism.
- 22. What is Cytochrome P-450? Explain its function and mechanism of action.
- 23. What is cis-platin? Discuss its pharmaceutical application and the mechanism of action.
- 24. What is Na+/K+ pump? Explain the mechanism of its function.

 $(8 \times 12 = 16 \text{ weighatge})$

Part C

Answer any two questions.

Each question carries a weightage of 4.

- 25. Briefly explain the structure and function o Haemoglobin. Give the mechanism of oxygen transport by haemoglobin in living system.
- 26. What are LNCCs and HNCCs? Discuss the polyhedral skeletal electron pair approach (Mingo's rules). Find out the total electron count, polyhedral electron count and predict the structures of [Os₅ (CO)₁₅]²⁻ and Ru₅C (CO)₁₆.
- 27. Explain the mechanism of Wacker process using catalytic cycle. Give evidences for the mechanism.
- 28. What is Wilkinson catalyst? Give the industrial importance of this catalyst. Briefly discuss the mechanism of the reaction. How the nature of the alkene affect the rate of the reaction?

 $(2 \times 4 = 8 \text{ weightage})$