THIRD SEMESTER B.Sc. DEGREE EXAMINATION, NOVEMBER 2015

(CUCBCSS—UG)

Complementary Course

CHE 3C 03—ORGANIC CHEMISTRY

Time : Three Hours-

Maximum: 64 Marks

Section A (One Word Answer)

Answer **all** questions. Each question carries 1 mark.

1. The self linking property of Carbon is known as _____

2. The type of hybridization of Carbon in methyl radical is _____

3. The optical isomers which are mirror images of each other are called

4. Among geometrical isomers of But-2-ene-1,4-dioic acid, the isomer having zero dipole moment is

5. One example for meta- orientative substituent is ———

6. The electrophile in Sulphonation reaction is _____

. ______ is a **pyrimidine** base present in RNA.

8. The **zwitter** ion form of **glycine** is ______

9. Oils and Fats are <u>of higher fatty ac</u>ids.

10. Give one example for an essential oil.

(10 x 1 = 10 marks)

Section B (Short Answer)

Answer any seven questions. Each question carries 2 marks.

- 11. Draw the structure of geometrical isomers of But-2-ene.
- 12. Discuss briefly on isomerism in **disubstituted** benzene compounds.
- 13. What is Huckel's rule ? Explain the aromaticity of Tropylium cation using it.
- 14. What are the products obtained when benzene is first chlorinated and then nitrated ? Justify your answer.
- 15. How alcohols can be prepared by using Grignard reagent ? Explain.

- 16. Write briefly on Williamsons's ether synthesis with one example.
- 17. What is meant by denaturation of protein?
- 18. What is the pentose sugar present in RNA? Draw its structure.
- 19. What is Iodine number of an oil ? What is its significance ?
- 20. What is meant by vulcanization ? Mention two advantages of vulcanized rubber.

(7 x 2 = 14 marks)

Section C (Paragraph Answer)

Answer any four questions. Each question carries 5 marks.

- 21. Taking suitable examples compare the acidity of aliphatic carboxylic acids.
- 22. Discuss the optical isomerism in Lactic acid. What is meant by resolution?
- 23. Explain the mechanism of nitration and Friedel Craft's reaction in benzene.
- 24. Write the mechanism of SN⁻ reactions of alkyl halides with one example.
- 25. Give any four synthetic applications of Benzene diazonium chloride.
- 26. Write a brief note on double helical structure of DNA.

 $(4 \times 5 = 20 \text{ marks})$

Section D (Essay)

Answer any two questions. Each question carries 10 marks.

- 27. (a) What is hyper conjugation ? How it can be used to explain extra stability of 2-Butene than 1-Butene.
 - (b) What are Carbocations? Discuss the relative stabilities of Carbocations.

(5 + 5 = 10 marks)

- 28. a What is Haloform reaction ? How will you distinguish between methanol and ethanol using lodoform test ?
 - (b) What is Lucas Test ? How will you distinguish primary, secondary and teriary alcohols by Luca's Test ?

(5 + 5 = 10 marks)

29. (a) Write short notes on :

- (i) Hofmann's bromamide reaction. and
- (ii) Hofmann's Carbylamine reaction.

(b) Compare the basicity of ammonia, methylamine and aniline.

(5 + = 10 marks)

30. (a) How are proteins classified based on amino acid residue?

- (b) Write any two examples for Enzymes and mention any two characteristics of enzymes.
- (c) Discuss primary, secondary and tertiary structure of proteins.

(3+2+5=10 marks)

[2 x 10 = 20 marks]