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# FIRST SEMESTER B.Sc. DEGREE EXAMINATION, JANUARY 2012

(CCSS)

## **Biochemistry—Complementary Course**

## BCIC 01—ELEMENTARY BIOCHEMISTRY

Time: Three Hours Maximum: 30 Weightage

I. Answer all twelve questions:	
1 The particles of dispersed phase in colloidal solution are called —	
2 The acid-base balancing solution is called	
3 What is the function of synovial fluid?	
4 Osmotic pressure depends on:	
(a) Temperature;	
(b) Solute Concentration;	
(c) Atmospheric pressure.	
5 Give an example of buffering system of blood.	
6 is an example of emulsifying agent.	
7 What is TLC?	
8 is the protein digesting enzyme of gastric juice.	
9 is the polysaccharide of Saliva.	
10 In a pH scale the acidic range lies below the pH	
11 The finely divided oil droplets in water is called	
12 Loss of atoms or groups from adjacent carbon atoms is called 1	
(a) Decarboxylation.	
(b) Addition reaction.	
(c) Elimination reaction.	
	$(12 \times \frac{1}{4} = 3 \text{ weightage})$
II. Answer all nine questions:	
13 What is Bronstead's definition of acids and bases?	
14 What is Osmosis?	

Turn over

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- 15 What is Lyophilic colloid?
- 1.6 What is Oxidation and reduction reaction?
- 1.7 Define molality of a solution:
- 1.8 Define buffering capacity.
- 19 What is optical activity?
- 20 What is the meaning of pKa?
- 21 What is a decarboxylation reaction? Give an example.

 $(9 \times 1 = 9 \text{ weightage})$ 

#### III. Answer any five questions:

- 22 Write a note on the normal constituents of urine.
- 23 Describe a method of pH determination.
- 24 Write an account of addition and substitution reaction with example.
- 25 Write a short note on dialysis.
- 26 Give the principle of colorimetry.
- 27 Explain the phenomenon of Brownian movement.
- 28 Water is an amphoteric molecule—Discuss.

 $(5 \times 2 = 10 \text{ weightage})$ 

### IV. Answer any two questions:

- 29 Describe the principles and application of electrophoresis in the separation of natural products.
- 30 Write an account of the function and composition of saliva, gastric juice and bile.
- 31 Explain the biochemistry of blood clotting.

 $(2 \times 4 = 8 \text{ weightage})$