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Name

FOURTH SEMESTER B.Sc. DEGREE EXAMINATION, APRIL/MAY 2015

(U.G.-CCSS)

Core Course-Chemistry

CH 4B 07—ORGANIC CHEMISTRY – I

Time : Three Hours

Maximum : 30 Weightage

Write equations wherever necessary.

Section A

- I. Multiple choice and fill in the blanks type questions. Answer all *twelve* questions. Each question carries a weightage of 1/4 :
 - 1 The type of hybridisation of carbon in ethane is :

(a) SP.	(b) SP ² .	
		~

- (c) SP^{\cdot}. (d) SP and SP².
- 2 Which of the following hydrocarbon is obtained by the Wurtz reaction between CH₃Br and C,H₃Br with metallic sodium in ether medium ?

(a) Ethane.	(b) Propane.
(c) Butane.	(d) All these.

3 The cyclo alkane which is not expected to have ring strain is :

- (a) Cyclopropane. (b) Cyclobutane.
- (c) Cyclohexane. (d) Cyclopentane.

Which of the following compound will exhibit geometrical isomerism?

- (a) Butane. (b) 2-butyne.
- (c) 2-butene. (d) All these.

long the **carbo** cations **Benzyl carbo** cation (A), Ally! **Carbo** cation **(B)** and a secondary **carbo** cation (C), the order of the stability is :

- (a) A > B > C. (b) B > A > C.
- (c) C > B > A. (d) C > A > B.

The molecule which exhibits optical isomerism is :

- (a) Isobutyl chloride. (b) Sec. butyl chloride.
- (c) Tert. butyl chloride. (d) n-butyl chloride.

Turn over

7 Calcium carbide on hydrolysis gives ____

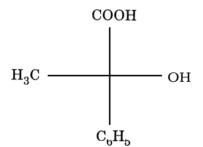
8 2-butyne on reduction with sodium in liquid ammonia gives _____

- 9 Natural rubber is a polymer of _____
- 10 Optical isomers which are mirror images of each other are known as _____
- 11 An example of a meta directing group is _____
- 12 Nitration of aromatic compound is an example of ______ substitution.

 $(12 \text{ x} \frac{1}{4} = 3 \text{ weightage})$

Section **B**

- II. Short Answer type questions. Answer all *nine* questions. Each question carries a weightage of 1.13 What is resonance ? Explain with an example.
 - 14 What is Kolbe reaction?
 - 15 How is polyethylene prepared?
 - 16 Draw the structure of geraniol.
 - 17 What is hydroboration?
 - 18 Draw the D and L forms of Erythrose.
 - 19 Assign the absolute configuration (R or S) of the molecule



- 20 What is meant by C is hydroxylation?
- 21 Write any two characteristics of enantiomers.

 $(9 \ge 1 = 9 \text{ weightage})$

Section C

- **III.** Short essays *or* paragraph questions. Answer any *five* questions. Each question carries a weightage of 2 :
 - 22 What is Corey-House reaction ?
 - 23 Explain Steric effect with one example.
 - 24 How will you prepare 1-butyne from acetylene ?
 - 25 Discuss the E and Z designation of geometrical isomers.
 - 26 What are addition polymers ? How are teflon polymers prepared ?

28 Write briefly on asymmetric synthesis.

 $(5 \times 2 = 10 \text{ weightage})$

Section. D

- IV. Essay questions. Answer any two questions. Each question carries a weightage of 4 :
 - 29 (i) Outline Bayer's strain theory. Calculate the angle strain for various alicyclic compounds and predict their relative stability.
 - (ii) Write a note on acidity of alkynes.
 - 30 (i) What are free radicals? Give examples.
 - (ii) Write a short note on the stability of free radicals.
 - 31 (i) Define aromaticity and state Huckel's rule.
 - (ii) Discuss the structure and stability of benzene.

(2 x 4 = 8 weightage)