D 53528

Name.....

Reg. No.....

# FIRST SEMESTER B.Sc. DEGREE EXAMINATION, JANUARY 2014

## (U.G.-CCSS)

#### **Complementary Course**

### **Physics**

### PH IC 01—PROPERTIES OF MATTER AND THERMODYNAMICS

#### (2013 Admission onwards)

Time : Three Hours

Maximum : 30 Weightage

### Part A

Answer **all** questions. Each question carries 1/4 weightage.

1. There is no change in volume of a wire due to change in its length on stretching. The Poisson ratio of the wire is :

(a) $-0.5$ .	(b) 0.5.
(c) <b>O</b> .	(d) 1.

2. The bulk modulus for an incompressible liquid is :

(a) Zero.	(b) One.
$() 1011 N m^{2}$	

(c) $10^{11}$ N-m.	(d) Infinite.
(c) IO $M=m$ .	(d) Infinite

<sup>3.</sup> The rigidity modulus of a material in the form of a wire can be determined using \_\_\_\_\_

4. With rise of temperature the surface tension of a liquid :

(a) Does not change.	(b) Increases.
4 N -	

- (c) Decreases. (d) Becomes zero.
- 5. The velocity of a falling raindrop attains limited value due to :
  - (a) Air current. (b) Up thrust of air.
  - (c) Surface tension. (d) Viscous force exerted by air.
- 6. Clouds float in the atmosphere because of :

(a) Low temperature.	(b) Low viscosity.
(c) Low density.	(d) Low surface tension.

- 7. The change in the internal energy of a gas is directly proportional to :
  - (a) Change in volume. (b) Change in pressure
  - (c) Change in temperature. (d) None of these.

Turn over

(Pages : 3)

8. The ratio of two specific heats of a diatonic gas is :

(a) 1.66.	(b) 1.4.
(c) 1.33.	(d) 1.21.

9. Change in entropy depends :

10.

11.

- (a) On the transfer of heat.
  (b) On change of temperature.
  (c) On the transfer of mass.
  (d) On the thermodynamic state.

  For a thermodynamic system work done in a process depends on :

  (a) The path.
  (b) State of the system.
  (c) External Pressure.
  (d) Temperature and Pressure.

  The efficiency of a heat engine working between reservoirs at temperature 327°C and 27°C is :

  (a) 25%.
  (b) 50%.
  - (c) 75%. (d) 100%.

<sup>12.</sup> The quantity remaining constant in the isothermal expansion of an ideal gas is

(a) Internal energy. (b) Heat.

(c) Pressure. (d) Temperature and Pressure.

 $(12 \text{ x} \frac{1}{4} = 3 \text{ weightage})$ 

#### Part B

Answer **all** questions. Each question carries 1 weightage.

- 13. What is meant by elastic hysteresis?
- 14. What is bulk modulus of elasticity ?
- <sup>15.</sup> Will you prefer a thin or thick handle to carry your bag. Why ?
- <sup>16.</sup> Distinguish between streamline flow and Turbulent flow of liquids.
- 17. What is the significance of Stoke's formula ? What is its use ?
- 18. How does temperature fall with height ?
- <sup>19</sup>. What are the limitations of the first law of thermodynamics ?
- 20. Give two conditions of obtaining maximum amount of work.
- 21. What is a reversible cycle ?

 $(9 \times 1 = 9 \text{ weightage})$ 

#### Part C

## Answer any **five** questions. Each question carries **2** weightage.

- 22. What is Stress energy ? A wire 4 m long and 3 x 10 m in diameter is stretched by a force of 8 kgwt. If the extension in the length amounts to 1.5 mm. Calculate the energy stored in the wire.
- 23. What is bending moment ? Derive an expression for the bending moment of a horizontal beam fixed at one end and loaded at the other end.
- 24. Calculate the work spend in spraying a drop of water of 1 mm radius into one million droplets of the same size. (S.T. of water = 0.0072N/m)
- 25. Calculate the mass of water flowing in 10 minutes through a tube 0.001 m in diameter, 0.4 m long under a constant pressure head of 20 cm of water. Coefficient of viscosity of water = 0.000089 SI units.
- 26. A tyre is pumped to a pressure of 2 atmospheres at 15°C when it suddenly bursts. Calculate the drop in temperature.
- 27. One gram of hydrogen occupies 11.1 litres at 0°C and 76 cm of mercury. What is the work done by the gas if heated to 1°C at constant pressure and how much heat must be supplied to it in the process ? Specific heat at constant volume is 2.411.
- 28. Show that the second law of thermodynamics enables us to define a scale of temperature independent of the properties of working substance. How is the scale realized in practice ?

 $(5 \times 2 = 10 \text{ weightage})$ 

#### Part D

## Answer any **two** questions. Each question carries **4** weightage.

- <sup>29.</sup> Describe with the theory the Torsion pendulum method of determining the rigidity modulus of a material in the form of a wire.
- 30. Describe Poiseuille's method of determining the coefficient of viscosity of a low viscous liquid. What are the factors on which viscosity of gases depend ?
- **31.** Explain the concept of reversible and irreversible process. Show that the efficiency of a reversible engine is maximum.

 $(2 \times 4 = 8 \text{ weightage})$