C 4758

(**Pages : 2**)

Name

Reg. No.

SECOND SEMESTER M.Sc. DEGREE EXAMINATION, JUNE 2016

(CUCSS)

Chemistry

CH 2C 08—ELECTROCHEMISTRY, SOLID-STATE CHEMISTRY AND STATISTICAL THERMODYNAMICS

(2015 Admissions)

Time : Three Hours

Maximum: 36 Weightage

Part A

Answer **all** questions. Each question carries a **weightage** of 1.

- 1. Write electrode reactions in the dry cell (Zn, MnO₂).
- 2. Write equation for the activity of the following electrolytes **in terms of molal concentration** and mean ionic activity coefficient :

(a) MX₂ ; (b) $M_2 X_3$.

Define exchange current density. Explain its significance.

- 4. .Explain the significance of slope and intercept of a Tafol plot.
- 5. Write Hermann-Mauguin symbol for the following (a) \mathbf{D}_{ad} ; (b) \mathbf{C}_{4v} .
- 6. Explain the term "glide plane".
- 7. Define Fermi level. Explain its significance.
- 8. What is birefringence ? Explain.
- 9. How many ways you can distribute two particles among three degenerate levels assuming (a) Bose Einstein statistics ; (b) Fermi Dirac statistics ?
- 10. Rationalise third law of thermodynamics using statistical concepts.
- 11. Calculate the heat capacity of solid (with characteristic temperature of 1000 K) at 10 K.
- 12. What do you mean by dilute system?

(12 x 1 = 12)

Part B

Answer any **eight** questions. Each question carries a **weightage** of 2.

- 13. Calculate the mean ionic activity coefficient of 0.01 molal LaCl₃ in water at 25° C. A = 0.509.
- 14. Explain the working of a lead acid battery.

Turn over

- 15. Write a brief account of the various models of electrical double layer.
- 16. Briefly explain one of the theories of hydrogen over voltage.
- 17. Draw stereographic projection for (222) system.
- 18. Write briefly on the application of non-stoichiometric compounds.
- 19. Briefly explain Meisner effect.
- 20. Explain the working of a laser.
- 21. Calculate the residual entropy of H₂O.
- 22. Evaluate translational partition function of CO_2 at 0°C. and 1 atm. pressure.
- 23. Derive an equation for the vibrational contribution towards heat capacity of gases.
- 24. Briefly explain Bose-Einstein condensation.

 $(8 \times 2 = 16)$

Part C

Answer any two questions. Each question carries a weightage of 4.

- 25. What are the assumptions in **Debye-Hückel theory**? Following the theory, derive **Debye Hückel** limiting law.
- 26. Derive Butler-Volmer equation.
- 27. Derive Maxwell Boltzman statistics. Discuss.
- 28. Apply Fermi Dirac statistics for electrons in metals. Discuss.

 $(2 \times 4 = 8)$