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SIXTH SEMESTER B.Sc. DEGREE EXAMINATION, MARCH 2017

(CUCBCSS—UG)

Chemistry

CHE 6B 13 (E2)—POLYMER CHEMISTRY

Time: Three Hours

Maximum: 80 Marks

Part A

Answer all questions. Each carries 1 mark.

- 1. Give the structure of PET.
- 2. Name one plasticizer for PVC.
- 3. Give the structure of polybutadiene.
- 4. What are the monomers of Phenolic Resin?
- 5. What is photodegradation?
- 6. What is Kevlar?
- 7. Give the applications of Polyurethanes?
- 8. What are the monomers of butyl rubber?
- 9. What is gelation?
- 10. How can you determine the crystallinity of a polymer?

 $(10 \times 1 = 10 \text{ marks})$

Part B

Answer any ten questions. Each carries 2 marks.

- 11. What is ring opening polymerisation?
- 12. What is nitrile rubber, give the uses.
- 13. Give Mark Houwink equation.
- 14. What are heat resistant polymers?
- 15. Distinguish between Tg and Tm?
- 16. Write on natural polymers.
- 17. What is tacticity?

Turn over

- 18. Explain ring opening polymerisation.
- 19. Distinguish between thermoplastics and thermosettings.
- 20. What is emulsion polymerisation?
- 21. What is meant by crosslinking?
- 22. Give the synthetic applications of phenolic resins.

 $(10 \times 2 = 20 \text{ marks})$

Part C

Answer any **five** questions. Each carries 6 marks.

- 23. What are sedimentation and viscosity average molecular weights?
- 24. What is polydispersity and molecular weight distribution?
- 25. What are chain transfer reactions?
- 26. Represent isotactic and syndiotactic polymers.
- 27. How will you distinguish between plastics, fibres and elastomers?
- 28. Explain injection moulding with diagram.
- 29. What is film extrusion?
- 30. Discuss end group analysis method for molecular weight determination.

 $(5 \times 6 = 30 \text{ marks})$

Part D

Answer any two questions. Each carries 10 marks.

- 31. Compare step reaction and chain reaction polymerisations.
- 32. Explain different polymerisation techniques.
- 33. Give the preparation, properties and uses of PMMA.
- 34. What are sedimentation and viscosity average molecular weights?

 $(2 \times 10 = 20 \text{ marks})$